Erratum to: The influences of filament temperature on the structure of boron nitride films and its tribological characterization for microforming die application

Yong Jin¹*, Shigeo Yasuhara², Tetsuhide Shimizu¹, and Ming Yang¹

¹ Graduate School of System Design, Tokyo Metropolitan University, Hino 191-0065, Japan
² Japan Advanced Chemicals, 3131-4 Tana, Chuo-ku, Segamihara-shi, Kanagawa 252-0244, Japan

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We realized that due to an unfortunate mistake Figure 2 was wrong in page 3 of the paper. Following the “Erratum”, we publish the article with the correct version of Figure 2. We apologize for the inconvenience that this mistake may have caused.
The influences of filament temperature on the structure of boron nitride films and its tribological characterization for microforming die application

Yong Jin¹,*, Shigeo Yasuhara², Tetsuhide Shimizu¹, and Ming Yang¹

¹ Graduate School of System Design, Tokyo Metropolitan University, Hino 191-0065, Japan
² Japan Advanced Chemicals, 3131-4 Tana, Chuo-ku, Segamihara-shi, Kanagawa 252-0244, Japan

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Abstract – Boron nitride film was deposited on (1 0 0)-oriented silicon substrate by hot filament assisted chemical vapor deposition. The B[N(CH₃)₂]₃ (Tris(dimethylamino)borane, TDMAB) was used as the single source precursor both for boron and nitride, and ammonia gas was used as the extra source to increase the N concentration in the films. Elemental composition of the films deposited under different filament temperatures were measured by energy dispersive X-ray (EDX) analysis, and the structure of the films were measured by X-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FT-IR). The boron nitride films deposited under lower filament temperature was amorphous, while BN films contain hexagonal structure were deposited at higher filament temperature. To verify whether the films can be applied for microforming die, ball on disk test was carried out using pure titanium ball as the counterpart to investigate the interfacial behavior between the films and pure titanium. The results show that the reaction between the films and pure titanium was low as there was no titanium adhesion on the wear track when the film was remained.

Key words: Filament temperature, Boron nitride, Structure, Tribological, Microforming

1. Introduction

Because of the similar crystal structure and properties as diamond, cubic boron nitride (cBN) has gained more and more attention from the last decades. And some properties of cBN are superior to diamond, like high temperature oxidation resistance and chemical inertness to ferrous alloys. Due to these properties, cBN has a wide application field as cutting tools and wear resistant coatings used in metal forming [1].

Various techniques were used to deposit cBN films, such as physical vapor deposition (PVD) and chemical vapor deposition (CVD) methods. These methods like ion-beam assisted deposition [2], RF sputtering [3] and ion-assisted CVD [4] have been succeeded in depositing nano-cBN films. However, these methods involve energetic ion bombardment, which causes the high compressive stress in the films. This internal stress leads the films cracking and delamination when the films thickness exceeds to several hundreds of nanometers, and it is difficult to deposit thick cBN films over 1 μm without delamination.

Compared with the PVD methods or plasma assisted CVD methods with high stress in the films, a simple chemical way for deposition of thick cBN films would be of high interest for industrial applications because of reducing the stress in the film. Weissenbacher and Haubner [5] used hot filament CVD to decompose the Triethylborazine (C₆H₁₈B₃N₃) source on various substrates, and obtained the B, C and N containing layers under different deposition conditions. Dumont et al. [6] used the Tris(dimethylamino)borane (TDMAB) as the single source precursor, and deposited the BN films on Si substrate under different deposition temperature. However, some difficulties appear in the deposition of cBN film by CVD without energetic plasma bombardment, and there has no available thermodynamic data as well as the exact transition temperature between hBN and cBN.

In this work, BN films were deposited by hot filament assisted CVD method, and the TDMAB gas was used as the single source precursor which was decomposed and activated by the hot filament. The influence of different filament
temperature on the structure of BN film was studied. On the other hand, in order to investigate the possibility of the films applied to microforming die for forming pure titanium, the tribological properties and interfacial behavior between the films and pure titanium were investigated by using ball on disk test.

2. Experiment

BN films were deposited on (1 0 0)-oriented silicon substrate by hot filament assisted CVD. Figure 1 shows the schematic diagram of the experimental set up which was designed by Japan Advanced Chemicals Ltd. The B[N(CH3)2]3 (TDMAB) gas was chose as the single source precursor since its nontoxic properties and the fixed ratio of boron and nitrogen. BN films were deposited under different filament temperatures which were controlled by a direct-current supply power. Since the N content in the films were much lower than the B content, ammonia gas was used as the extra gas to increases the N content in the films. Si substrates were cleaned by acetone and ethanol in an ultrasonic bath before deposition, and put on a graphite substrate which was heated to 700 °C by a SiC ceramic heater. The detail deposition conditions are summarized in Table 1.

All the films obtained were analyzed by energy dispersive X-ray (EDX) analysis to estimate the elemental compositions. X-ray diffraction (XRD) and Fourier transform infrared spectroscopy (FT-IR) were used to identify the structure of the films. The thickness of the film was measured by atomic force microscopy (AFM). The interfacial behavior and friction coefficient of the films were measured by ball-on-disk (BOD) tribometer. A constant normal force of 1 N was applied for all the tests, and the force corresponds to an initial Hertzian mean contact pressure of about 343 MPa [7]. The details of BOD test conditions are summarized in Table 2.

3. Results and discussion

3.1. Measurement of the BN films

3.1.1. Elemental composition and structure of the films

Table 3 shows the elemental composition of the films deposited with different filament temperature measured by EDX analysis. The elemental composition of the films was influenced by the different filament temperature. With increasing the filament temperature, the N concentration in the films increased. There is no C contains in the films when the filament temperature increased to 1800 °C. The thicknesses of the films deposited under different filament temperatures were 133 nm, 265 nm and 316 nm, respectively. This suggested that the higher temperature of filament makes the precursor being decomposed completely, and the deposition rate of the films were increased with increasing the filament temperature.

The structures of the films were analyzed by FT-IR and XRD analysis. The FT-IR analysis is a most important and widely used tool for characterizing BN films, for hBN, the two IR-active phonons have TO frequencies at about ~780 and ~1370 cm⁻¹, respectively. The 1370 cm⁻¹ mode is a stretching of the B-N bond within the basal plane, and the 780 cm⁻¹ mode is a bending of the B-N-B bond between the basal planes [1]. Figure 2 shows the FT-IR results of the films as shown in Table 3. The FT-IR spectra of the films deposited with different filament temperature were different. At the lower filament temperature, no clear peak was observed, the spectra looks smooth and with some broad peaks. When the filament temperature increased to 1800 °C, two peaks with wide wavenumber were detected. The intensity of the peaks increase and become narrow when the filament temperature increased to 2000 °C. These two peaks correspond well to the peaks of hBN in the FT-IR spectra [1]. It indicated that
the film deposited under lower filament temperature has disordered structures or amorphous, and hBN films could be deposited under higher filament temperatures.

Figure 3 shows the XRD results of the films as shown in Table 3. It can be found that no peak was detected in the XRD spectra when the film was deposited under lower filament temperature. As the filament temperature increased to 1800 °C, a peak around 23° was observed, and the intensity of this peak increases as the filament temperature further increased. It indicated that the crystallization degree of the films was increased with increasing the filament temperature. However, the (0 0 2) diffraction peak position for hBN is reported at 2θ = 26.8° [8], the peak shift to the small angle may be caused by the rich B content in the film. The excessive B atom may incorporate into the structure, and causes the lattice constant increase. According to Bragg’s low [9], the angle
would be decreased if the lattice constant increase. This needs further investigation in the future.

From the FT-IR and XRD measurements, it suggested that the hBN films were deposited at higher filament temperature, and the crystallization degree of the BN films become better as the filament temperature increased. In order to investigate whether the hBN films can be applied for microforming die for forming pure titanium, the BOD tests were carried out to investigate the tribological properties and interfacial behavior between the BN films and pure titanium.

3.1.2. Tribological properties and interfacial behavior of the film

Figure 4 shows the friction coefficient of the films deposited with different filament temperature. From these results, it can be observed that the films deposited at lower filament temperatures show the higher and big vibration friction coefficient than the film deposited at higher filament temperature. The friction coefficient of the film deposited at higher filament temperature was different compare with the films deposited at lower filament temperature. At the initial stage of the BOD test, the friction coefficient is low and smooth, while it was increased rapidly and vibrated after 20 laps. After BOD test, many wear debris were found on the edge of the wear track. As the hBN film was very soft and it was easily to be removed, the wear debris on the wear track may increase the friction coefficient, and the friction coefficient increased after the film was worn out.

As the life time of the film was too short due to the thin thickness, in order to investigate the wear properties and interfacial behavior of the film sliding against with pure titanium, the thick hBN film was needed. Figure 5 shows the surface morphology of the films deposited for 120 min with the filament temperature at 2000 °C, the other deposition conditions were the same as before. The thickness of the film is around 2 μm measured by AFM. From the SEM image of the hBN film, it was found that the film has regular structures with the size around 1 μm.

Figure 6 shows the friction coefficient of the film measured by BOD test under the same testing condition except the sliding lap increased from 100 to 800. The friction coefficient of the film shows the same tendency as the film with thin thickness. The friction coefficient of the film was low and stable at the initial stage, then it increases as the sliding laps increased. In order to study the interfacial behavior between the film and pure titanium, the elemental composition of the wear track on the film was measured during the initial stage and after the BOD test.

Figure 7 shows the surface morphology and EDX results of the film after 50 and 800 laps. The results show that there was no titanium adhesion on the wear track after 50 laps, when the sliding lap increased to 800 laps, the titanium was detected and adhesion on the wear track. It was considered that the titanium was adhesion on the film only when the film was worn out. In other words, if hBN film was remained, the titanium would not adhesion on the film.

From the BOD test, it was found that the hBN film has low friction coefficient and low reaction with pure titanium, it could be applied to the microforming die for forming pure titanium. However, the adhesion strength or the wear resistance of the film was low, it was easily to be worn out during the BOD test.

3.2. Discussion

BN films contain hexagonal structure could be deposited at higher filament temperature by the filament assisted chemical
vapor deposition, the higher temperature of filament makes the higher electron energy and lead to the hexagonal structure of BN formation. As there was no energetic ion/plasma bombardment during the deposition process, the internal stress in the films was much lower than the films deposited by PVD or PECVD process. The energy for cBN nucleation is higher than the energy for hBN film formation, if the electron energy from filament can be higher enough for cBN nucleation, it was possible to deposit the thick cBN film with lower internal stress by the filament assisted CVD process.

For microforming die application, the coating system should have low friction coefficient, low wear rate, high adhesion strength on the tool and low reaction with forming...
materials. The BOD test shows that the hBN films has low friction coefficient and chemical inertness to pure titanium, it could be used as the coatings for microforming tool to prevent pure titanium sticking on the tool surface. However, the wear rate of the hBN films was high and it was easily to be worn out. In order to apply the BN film to microforming die, future work should be done to decrease the wear rate of the film. As is known to all, the cubic boron nitride film has high hardness and low wear rate, if the hBN film contains some cubic structure, the wear resistance of the coating system could be improved. On the other hands, some other substrate material like WC should be used, as this material is a common material for tool application.

4. Conclusion

The BN film was deposited by hot filament CVD. The filament temperature influenced the decomposition rate of single source precursor and lead to different film's elemental composition and structure. The film deposited under lower filament temperature was amorphous, hBN film was deposited under higher filament temperature, and the crystallization degree of the film was improved at the higher filament temperature. The hBN film shows the lower and stable friction coefficient at the initial stage, moreover, there was no titanium adhesion on the film after ball on disk test when the film was remained. It was expected that the hBN film could be used for microforming die to prevent pure titanium adhesion on the tool surface.

References


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